Challenging in PBL Education on the Field of Chemistry, Machine and Mathematics

Yoshiyuki Uruma^{*}, Takeshi Gonda^{**}and Masaki Okugumo^{***}

*Department of Materials Science, National Institute of Technology, Yonago College, Hikona-cho, 4448 Yonago City, Japan

**Department of Mechanical Engineering, National Institute of Technology, Yonago College, Hikona-cho, 4448 Yonago City, Japan

***Department of Electrical and Computer Engineering, National Institute of Technology, Yonago College, Hikona-cho, 4448 Yonago City, Japan

> *E-Mail <u>uruma@yonago-k.ac.jp</u> **E-Mail t-gonda@yonago-k.ac.jp ***E-Mail okugumo@yonago-k.ac.jp

Abstract

It's important for teacher to educate the skill to students. Fundamentally ability is also important, but in recent years companies are focus on the ability to solve problem in cooperation with others. The class so far were the style that the teacher talks unilaterally methods. Base on the method, students who can understand contents of the lesson are no problem, but students who cannot understand one has problems. To solve this problem, it has been attention to active learning. This method is defined as interactive learning. Therefore the students have to participate more actively. As a result, the students are able to obtain a skill solving a problem and to solve fundamental problem.

Herein, we described chemistry, mechanical and mathematics subjects. From student's comments, this method clarified a useful for problem-solving ability.

Keywords: *PBL* education, Active Learning, NMR analysis, Open CAE, Group work

1. Introduction

Up to now, Kosen education has been conducting in order to bring up technical engineers. Therefore, Alumnae and alumni play an active role as efficient worker in the industrial society. However, shift with change in academic development, educational method need to be changed. Most of the teaching centres, schools, colleges, universities and tuition have their teaching methods based on lectures where instructors would lead the whole learning environment and convey information to students via power point screen or notes written on boards. This teaching strategy would give effects when students are interested and passionate on the learning process or instructors able to give consecutive motivation on their means of teaching. However, with today's student, lecturing does not hold their attention long. Students today have grown up dealing with interactive tools like Internet, computers and smartphones and these gadgets and media allow students to enjoy the real-world problem solving and also give opportunity to express their own views and hear their own voices. Educational life should work the same. A very good example of this is the projective base learning (PBL) engaged in the robotics projects. Students get into group to seek solution to real challenging problems. This process creates a learning community where all participants take responsibility for learning and achieving understanding of concepts by themselves, also deepens the communication skills. With this supportive evidence, implementing PBL in teaching of chemistry has come to action, yet not a major influence in education. In our college, students are allowed to engage in practical tasks which give a complete different learning exposure to chemistry, mechanics, and mathematics. In fact, they learn better and understand deeper through operating experimental procedures and discussion with their friends. This paper provides an overview of the findings in our department education system and discusses the consequences of implicating PBL based syllabus on educational development.

1-1. Outline of this experimental lesson

Our department of chemistry afford students with complete experience in understanding chemistry by variety of experiments based on biology, physical chemistry and analytic chemistry according to their education grade. Particularly, students of the fourth grade has more intense experimental tasks of 3hours a week. They are offered with opportunities to enhance their knowledge in chemistry through class includes 10 practical themes ; NMR spectroscopy (¹H NMR, ¹³C NMR) measurement, X-ray spectroscopy, density and viscosity against water-alcohol system, measurement of heat of solution for sodium thiosulfate, measurement of boiling point for methanol-benzene system, electromotive force of Daniel-Volta battery, hydrolysis rate of methyl acetate, logic circuit, absorption spectrum and emission spectrum, computer simulations.

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Figure 1. Syllabus of Experimental lesson (Japanese ver.)

The students are grouped and time is provided for students to do some research on their experiment, in they are assigned to design the own experiential procedures that aim to explain and describe the hypothesis of each task. This enable personalized learning. Participants set their own learning pace and concepts by merging the gap between theory and practice which play a crucial role retaining ideas gained from classroom learning. This journey also imparts high focus on collaboration and valuable discussion among members which increase students' engagement in chemistry Of course, teachers confirm the pre preparation by checking their notes and instructing them the safety alerts before they are allowed to carry out the experiment.

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Figure 2. Experimental notebook

1-2. Details of this experimental lesson

~ NMR Measurement of Unknown Organic Compounds~

NMR analysis is the most common and powerful tool used nowadays which provides scientists with substance or chemicals information of both stoichiometry and stereometry (Figure 3). Students are provided an unknown sample with known element constituents. The NMR data was measured by students. The data have to analyze until next lesson.



Figure 3. NMR equipment

Teacher taught a meaning chemical shift and split of signals (singlet, doublet and triplet). In the case of ethanol, the chemical composition formula weight is C2H6O1. Firstly Degree of unsaturated (D. U.) was calculated from chemical composition formula weight. D. U. shows a number of unsaturated bond and existence of cyclic structure in the molecule. As a result, a candidate molecular structure were easily obtained

(Figure 4). In this case this value show 0; unsaturated bond not involved in the molecule. From this knowledge, students can write two kinds of candidate structures (Figure 4).



Figure 4. Candidate structure for C2H6O1 Left: ethanol Light: diethyl ether

One out of these compounds determined by $^1\mathrm{H}$ NMR and $^{13}\mathrm{C}$ NMR analysis.

1-3. Results and Discussion



Figure 5. NMR spectra of ¹H NMR and ¹³C NMR

In the case of C2H6O1, it has two candidate structures (Figure 4). On the other hand a molecule has ethyl group and methoxy group. According to ¹H NMR

data, δ 1.11 ppm indicated triplet signal (3H) and δ 3.57 ppm also indicated quartet signal (2H). And ¹³C NMR spectrum shows two signals in the chart. As a result, there are two kinds of carbon atoms in the molecule. Finally it has been conclude an unknown sample as ethanol.

By thinking for a candidated structure and analyzing NMR data, students are going to have important skill.

The objective of this lesson is to be having ability for solving problem by discussing their friends.

How do students think about this lesson? I have been interested in student's thought. For some students, this experimental lesson may be pretty hard to write a report. However, a written report is good one. The NMR assignments of unknown samples were also written very well. I carried out the survey about how to proceed the experimental curriculum for the student. The survey includes following listed questions.

1) What grade are you in college?

- 2) How feel about this experimental curriculum?
- 3) If you feel good, please fill in your comment.
- 4) If you feel nothing or no good, please fill in your comment.

5) Please teach me about skills you are learning into the experiment.

These answers are below. The answer for question 2 shows much good feelings. Therefore answers for the question 3 are collecting a lot answers. Mainly answers are to be discussed with the student about not understanding problems and to be understood importance of team work. Answer for the question 4 is not answered. This is meaning that all students feel satisfactory. From the student's comments, they has been discussed with their friends about analysis of NMR data. According to discussion, students have an ability to solve problems for unknown matter and communication skills.

2-1. Introduction of OpenCAE software to experiments.

Recently, manufacturing development by CAE application is indispensable in research and development at companies.

Even at Kosen, there are few opportunities for students to learn numerical analysis while studying. I thought that experiencing numerical analysis at school was effective, and I took it into the class of student experiment.

On accounting to all the facts, the department of mechanical engineering in National Institute of Technology, Yonago Callege has offered a subject of "Mechanical Engineering Experiment Practicum" to 1st grade to 5th grade. In each grade, subjects that can

acquire 3 credits in 140 weeks per week, 3 weeks in all 30 weeks . In this subject, unique and high-tech contents (WHAT KNID OF UNIQUE AND HIGH_TECH CONTENT) are built in, so that students able to learn mechanical engineering thoroughly. For lower grade students (first to third grade), machining at the manufacturing center (practical factory) is the main practical task offered, and the experiment is conducted using materials such as metallographic observation and metallic materials test which are the fundamental chapters in dynamics system. For higher grades students (fourth to fifth grade), experiments about strength of thermodynamics, hydrodynamics materials, and mechanical mechanics are offered. For 5th grade student, we introduced the opportunity to set up "numerical simulation by finite element method" as a theme of strength of materials.

We decided to use ready-made software.

• Experiments must be 280 minutes lessons per theme.

(140 minutes per session, 1 theme setting in 2 weeks.)

· Computer proficiency for each students differ.

From the above, we decided to adopt "ADVENTURE on Windows (Windows version)" by the University of Tokyo ADVENTURE project in this theme. Figure 1 shows the basic screen of "ADVENTURE on Windows".



Figure 6 ADVENTURE on Windows

During the first class, we decided to introduce "ADVENTURE" for the following merit.

 \cdot It is open-source software and does not require expenses for purchasing software.

 \cdot The Windows version is much easier to set up than the Linux version.

 \cdot Since it runs on Windows, virtual software etc is unnecessary.

 \cdot Since "ADVENTURE" itself can be operated easily, it is easy to administrate in a short time.

2-2. How to use the software



Figure 7 Numerical Example (Cantilever beam)

In the numerical analysis of the example at the first week, everyone worked on the cantilever problem receiving the concentrated load as shown in Fig. 7, and the stress of the beams for the comparison of the numerical analysis result and the theoretical value of the flexural bending. In the second week, I focused on numerical analysis of tasks and report preparation.

The assignment was divided into 3 questions according to the degree of difficulty, and the students select one from those and answer in the report.

In this lesson, considering the difficulty level in numerical analysis, tasks 1 to 3 have the following contents.

Problem 1: When the H-shaped steel of Fig. 8 has a sectional shape in which the load is applied as the cantilevered beam of Fig.9.

Problem 2: When the H-shaped steel of Fig. 8 has a sectional shape in which the load is applied as the bothend supported beam in Fig. 10.

Problem 3: Stress analysis of a flat plate having a circular hole as shown in Fig. 11.



Figure 9 Exercises 1 (Cantilever beam)



Figure 10 Exercises 2 (Simply supported beam)



Figure 11 Exercises 3 (Flat plate with circular hole)

This class was introduced since 2017 and the deta of classes for two years has been accumulated.

Table1. Selectivity in exercise

Tack No	Selectivity			
Task INO.	2015	2016		
1	57 %	79 %		
2	0 %	0 %		
3	43 %	21 %		

3. Attempt of active learning in applied mathematics

In the 4th grade, we offer subjects aimed at applying mathematical knowledge such as vector analysis, complex analysis, Fourier analysis, Laplace transform, and probability statistics to other fields such as engineering.

In the active learning conducted this time, each question of periodic examination is allocated for each group, and each group learns and discusses so that other students can explain the allocated problems centered on leaders so that they can easily understand. After that, the leader carried out a board book and commentary on the blackboard and carried out at the class content such that teachers complemented and provided guidance at the appropriate time.

3-1. Assembling a lesson

The composition of the lesson is as follows.

Each group, the leader will explain in front of the students an easy-to-understand answer to the questions assigned by the leader.

Table 2 Contents of time table

Time(min)	Context
0~20	Introduction
20 ~ 35	Exercise(Group)
35 ~ 80	Presentation(Group)
80 ~ 90	Questionnaire

3-2. Exercise by group work

I asked the reader to create an example solution that other students could easily understand. I distributed A4 paper for each group and let me write the contents of the model answer and commentary on this paper.



Figure 12

Conclusions

In this paper, students carried out by the discussion with group member a structure analysis of organic compound. As a result, students are able to have the ability for solving the problem. This ability facilitate solution difficult problem and to create newly technology. In mechanics, although we have sampled few years. It can be confirmed that there is a large difference in the selectivity of problems. In both grade students, there is no student who chose task 2. From this result, task 3 has higher points and tips for the similar problem by using the Internet. Applied mathematics conducted group work by using active learing methods. From this result, the half students of this class answered 'it can be understood well'(Figure 12).

Acknowledgements

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